Experimental investigation of viruses and clay particles cotransport in unsaturated porous media

BolFate

EGU2014 HS8.1.7- 3470 **R308**



Vasiliki I. Syngouna ¹ and Constantinos V. Chrysikopoulos ²

Environmental Engineering Laboratory, Civil Engineering Department, University of Patras, Patras 26500, Greece (kikisygouna@upatras.gr) ² School of Environmental Engineering, Technical University of Crete, Chania 73100, Greece (cvc@enveng.tuc.gr)



Technical University

of Crete

Abstract

Suspended clay particles in groundwater can play a significant role as carriers of viruses, because, depending on the physicochemical conditions, clay particles may facilitate or hinder the mobility of viruses. This study examines the effects of clay colloids on the transport of viruses in variably saturated porous media. All cotransport experiments were conducted in partially saturated columns packed with glass beads, using bacteriophages MS2 and ΦX174 as model viruses, and kaolinite (KGa-1b) and montmorillonite (STx-1b) as model clay colloids. The various experimental collision efficiencies were determined using the classical colloid filtration theory. The experimental data indicated that the mass recovery of viruses and clay colloids decreased as the water saturation decreased. moments of the various breakthrough Temporal concentrations collected, suggested that the presence of clays significantly influenced virus transport and irreversible deposition onto glass beads. The mass recovery of both viruses, based on total effluent virus concentrations, was shown to reduce in the presence of suspended clay particles.

Materials and methods

Bacteriophages

MS2: an F-specific single-stranded RNA phage with effective particle diameter ranging from 24 to 26 nm ΦX174: a somatic single-stranded DNA phage with effective particle diameter ranging from 25 to 27 nm For the separation of viruses adsorbed onto clay particles from suspended viruses in the liquid phase, centrifugation was used as described in Syngouna and Chrysikopoulos (2013).

Clays

Kaolinite (KGa-1b): a well-crystallized kaolin from Washington County, Georgia

Montmorillonite (STx-1b): a Ca-rich montmorillonite, white, from Gonzales County, Texas

The <2 µm clay colloidal fraction was separated by sedimentation and then was purified (Rong et al., 2008)

Electrokinetic measurements

The zeta potentials were determined to be -40.4±3.7 mV for MS2, -31.78±1.25 mV for ФX174, -26.03±2.77 mV for KGa-1b, and -20.5±0.8 mV for STx-1b (Chrysikopoulos and Syngouna, 2012).

Glass beads 2mm

■ pH 7.0±0.2

Flow rate of Q=1.5 mL/min

Saturation level: 81-100%

Water potential: constant

In the column effluent:

 $C_{Total-v} = C_v + C_{vc}$

Experimental Set Up

- Plexiglass column Length 15.2 cm Internal diameter 2.6 cm
- Uniformly wet-packed

Cotransport Experiments

C_c: Suspended clay particles

C_{Total-v}: Total viruses C_v: Suspended viruses

C_{vc}: Viruses attached onto C_c

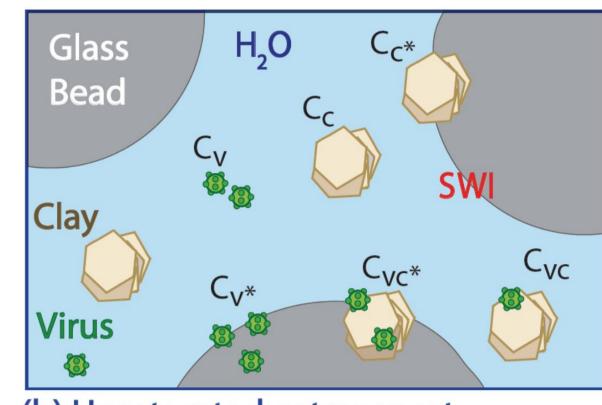
C_{c*}: Clays attached onto glass beads C_{v*}: Viruses attached onto glass beads

C_{vc*}: Viruses attached onto C_c*

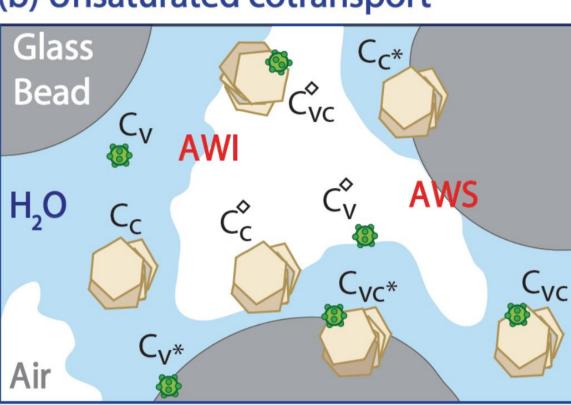
C_v♦: Viruses captured in AWI C_c: Clay colloids captured in AWI

C_{vc}♦: Viruses attached onto C_c♦

(a) Saturated cotransport



(b) Unsaturated cotransport



AWI: air-water interface AWS: air-water-solid **SWI**: solid-water interface

Colloid Filtration Theory

Experimental attachment efficiency (Kretzschmar et al., 1999):

$$\alpha_{\text{exp}} = -\frac{2}{3} \frac{d_{\text{c}}}{L(1-\theta_{\text{m}})\eta_{0}} \ln \left| \frac{C_{\text{ss}}}{C_{0}} \right|$$

where:

- θ_m [-] is the moisture content
- d_c [L] is the mean collector diameter
- C₀ [M/L³] is the influent colloid concentration
- C_{ss} [M/L³] is the effluent colloid concentration (steady state conditions)
- η_o is the single-collector contact efficiency (Tufenkjii and Elimelech, 2004)

Results and discussion

Transport experiments

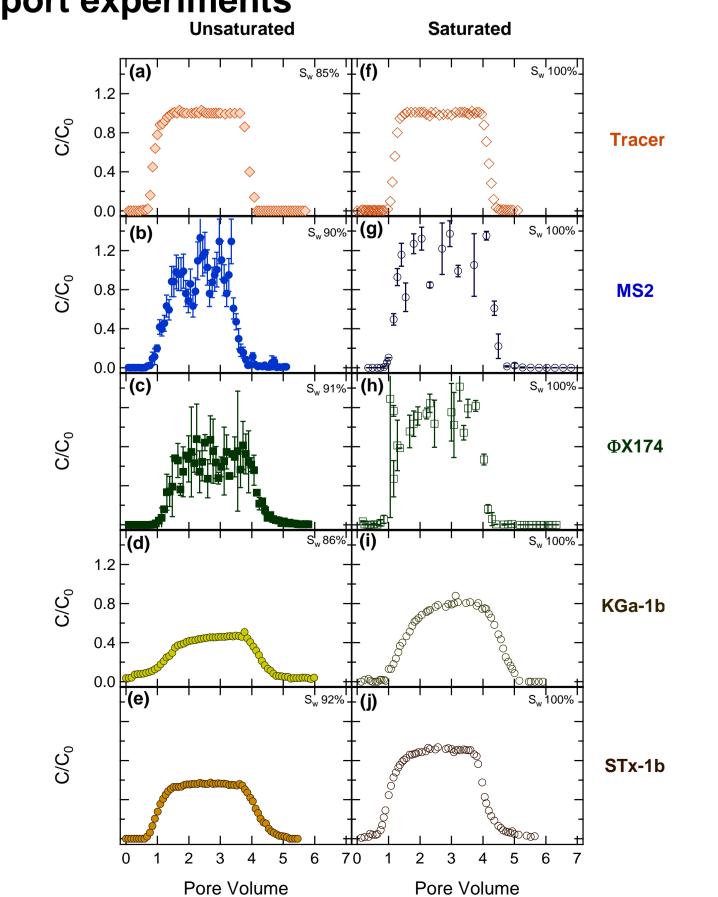


Figure 1. Experimental data of tracer, viruses (MS2, ФX174) and clays (KGa-1b, STx-1b) breakthrough in unsaturated (a,b,c,d,e) and saturated (f,g,h,i,j) columns packed with glass beads.

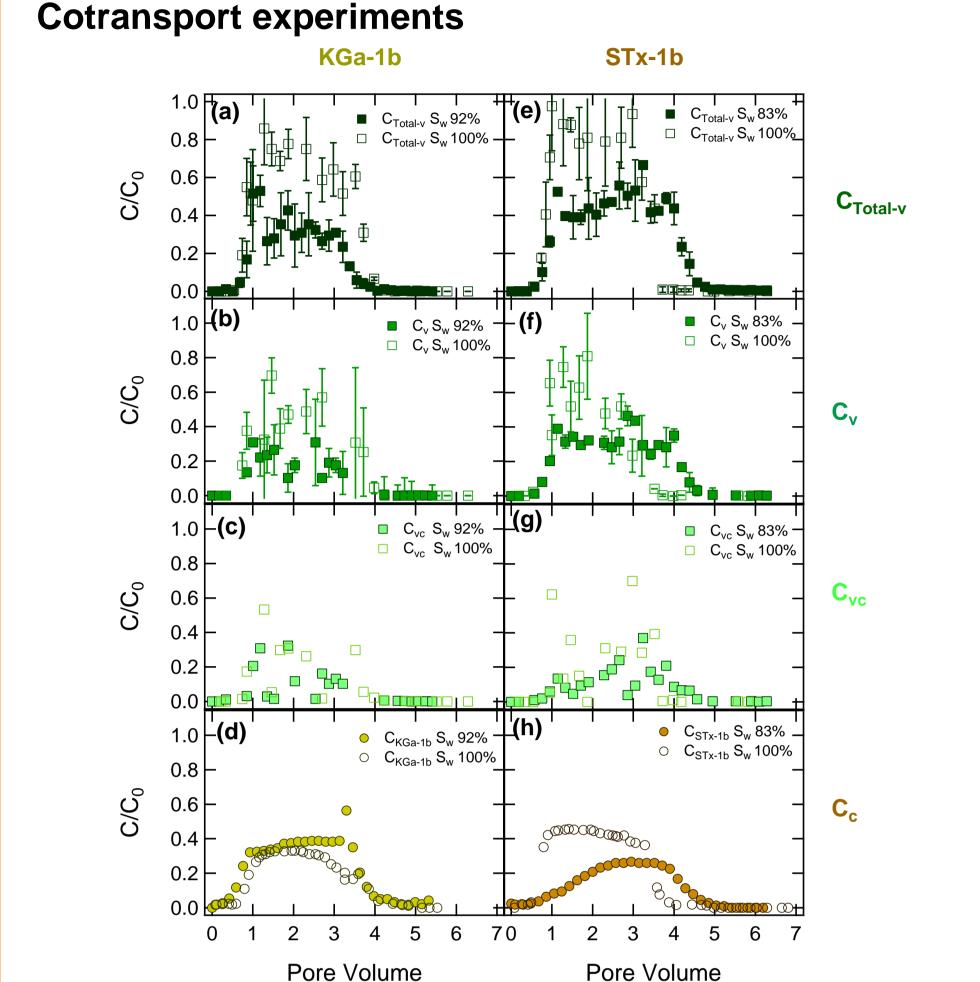


Figure 2. Experimental data for the cotransport of (a,b,c,d) ΦX174-KGa-1b and (e,f,g,h) ΦX174-STx-1b in both saturated (open symbols) and unsaturated (filled symbols) columns packed with glass beads.

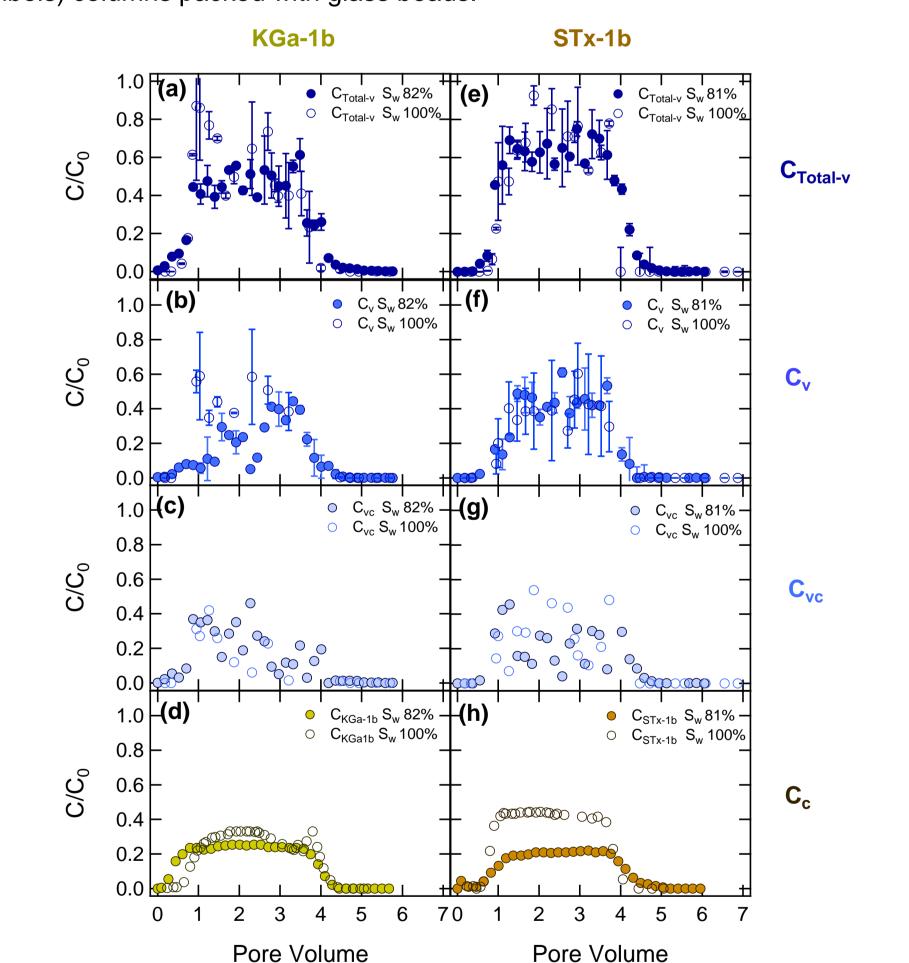


Figure 3. Experimental data for the cotransport of (a,b,c,d) MS2-KGa-1b and (e,f,g,h) MS2-STx-1b in both saturated (open symbols) and unsaturated (filled symbols) columns packed with glass beads.

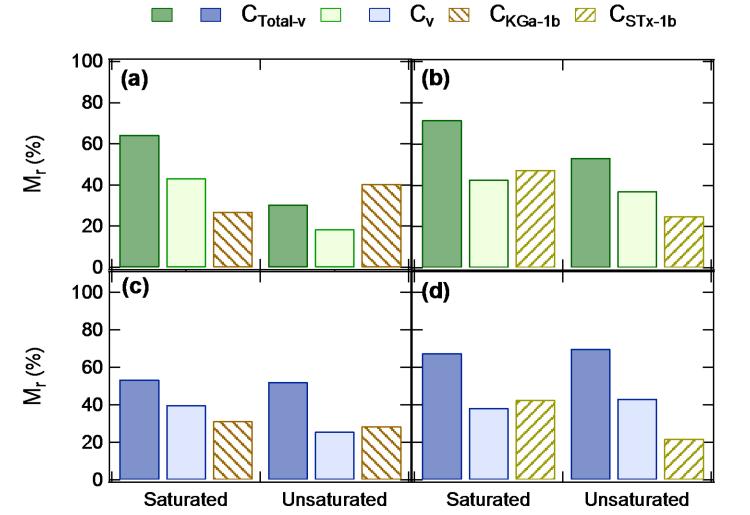
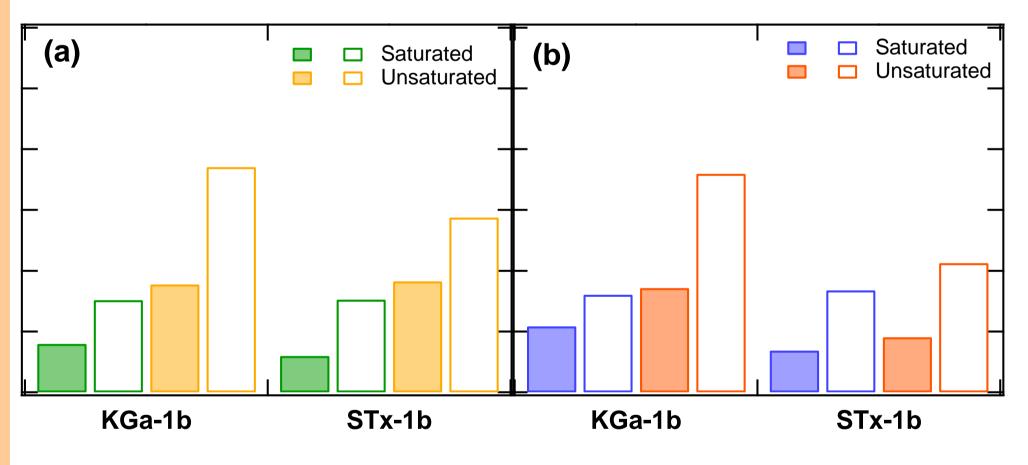


Figure 4. Calculated M_r values based on C_{Total-v} (solid columns), C_v (filled columns) and C_c (cross-shaded columns) for cotransport of: (a) ΦX174 with KGa-1b, (b) MS2 with KGa-1b, (c) ΦX174 with STx-1b, and (d) MS2 with STx-1b under saturated and unsaturated experimental conditions.

Table 1. Experimental conditions and estimated parameters for the cotransport experiments

Exp. No	Initial concentration C_{v0} , C_{c0}	θ _m (-)	S _w (-)	U (cm/min)	M _r (%) for C _{Total-v} or C _c	IVI _r (/0)	M _{1(i)} /M ₁ _(t) for C _{Total-v}	_{1(t)} for	αTotal–v	αν
Cotransport experiments ΦX174-KGa-1b										
1	2767 PFU/mL	-	1	0.74	64.55	43.41	0.82	0.83	0.080	0.152
	80.75 mg/L				27.2		0.87		0.648	
2	5100 PFU/mL	0.39	0.92	0.64	30.62	18.67	1.01	1.01	0.178	0.371
	75 mg/L				40.59		1.13		0.776	
			(DX174-ST	x-1b		L			
3	4817 PFU/mL	-	1	0.74	71.85	43	0.79	0.82	0.060	0.153
	122.21 mg/L				47.74		0.77		0.208	
4	7433 PFU/mL	0.37	0.83	0.72	53.52	37.32	1.11	1.08	0.183	0.288
	107 mg/L				25.11		1.18		0.597	
				MS2-KGa	·1b					
5	9767 PFU/mL	1	1	0.74	53.75	39.98	0.78	0.85	0.109	0.16
	76.98 mg/L				31.41		0.85		0.577	
6	21383 PFU/mL	0.37	0.83	0.72	52.22	25.79	1.08	1.20	0.172	0.360
	69 mg/L				28.64		1.01		0.930	0.172
				MS2-STx-	1b					
7	9500 PFU/mL	-	1	0.74	67.65	38.45	0.89	0.92	0.069	0.168
	92.56 mg/L				42.88		0.85		0.238	
8	11067 PFU/mL	0.35	0.81	0.7	69.98	43.42	1.09	1	0.091	0.213
	89 mg/L				22		1.12		0.612	



efficiencies $\alpha_{Total-v}$, based on $C_{Total-v}$ (filled columns) and $lpha_{_{
m VP}}$ (KGa-1b, STx-1b) and (b) MS2-clays (KGa-1b, cotransport

Conclusions

- The mass recovery of viruses and clay colloids decreased as the water saturation decreased.
- The mass recovery of both viruses was shown to reduce in the presence of suspended clay particles.
- Under saturated conditions, the transport of both C_{Total-v} and C_v was retarded, compared to the conservative tracer while under unsaturated conditions the opposite was observed.
- Under unsaturated conditions both clay particles hindered the transport of both viruses.
- In the presence of STx-1b, the $C_{vc} = C_{Total-v} C_v$ values of both viruses were higher than those in the presence of KGa-1b under both saturated and unsaturated conditions.
- In the presence of both KGa-1b and STx-1b, $\alpha_{Total-v}$ and α_v values increased with decreasing saturation level.
- References 1. V.I. Syngouna, C.V. Chrysikopoulos, Colloids Surf. A: Physicochem. Eng. Aspects 416 (2013)
- 56-65. 2. X. Rong, Q. Huanga, X. He, H. Chen, P. Cai, W. Liang, Colloids Surf. B: Biointerfaces 64
- (2008) 49-55.3. C.V. Chrysikopoulos, V.I. Syngouna. Colloids Surf. B: Biointerfaces. 92 (2012) 74-83.
- 4. R. Kretzschmar, M. Borkovec, D. Grolimund, M. Elimelech, Adv. Agron. 65 (1999) 121–193. 5. N. Tufenkji, M. Elimelech, Environ. Sci. Technol. 38 (2004) 529-536.



This research has been co-financed by the European Union (European Social Fund-ESF) and Greek national funds through the Operational program "Education and Lifelong Learning" of the National Strategic Reference Framework (NSRF)-Research Funding Programs: Heracleitus II and Aristeia I (No. 1185). Investing in knowledge society through the European Social Fund.